# Metals and their Compounds Lecture 3.4 

Face-centered cube (p 417 BLB)

cube length $=a \quad$ atom radius $=r$

Atoms touch along face-diagonal of cube, so

$$
\begin{gathered}
b=4 r \text { and } r=b / 4 \\
b^{2}=a^{2}+a^{2} \text { so } b^{2}=2 a^{2} \text { so } b=\sqrt{ } 2 a \\
\text { so } r=\sqrt{ } 2 a / 4=a /(2 \sqrt{ } 2)
\end{gathered}
$$

